

Verfahren zur Modifikation von FAME

Überblick über neuere Entwicklungen

Processes to modify FAME – Survey on recent developments



Axel Munack¹

Jürgen Krahl²

Michael Meier³

¹ Johann Heinrich von Thünen Institute

² Coburg University of Applied Sciences

³ University of Potsdam

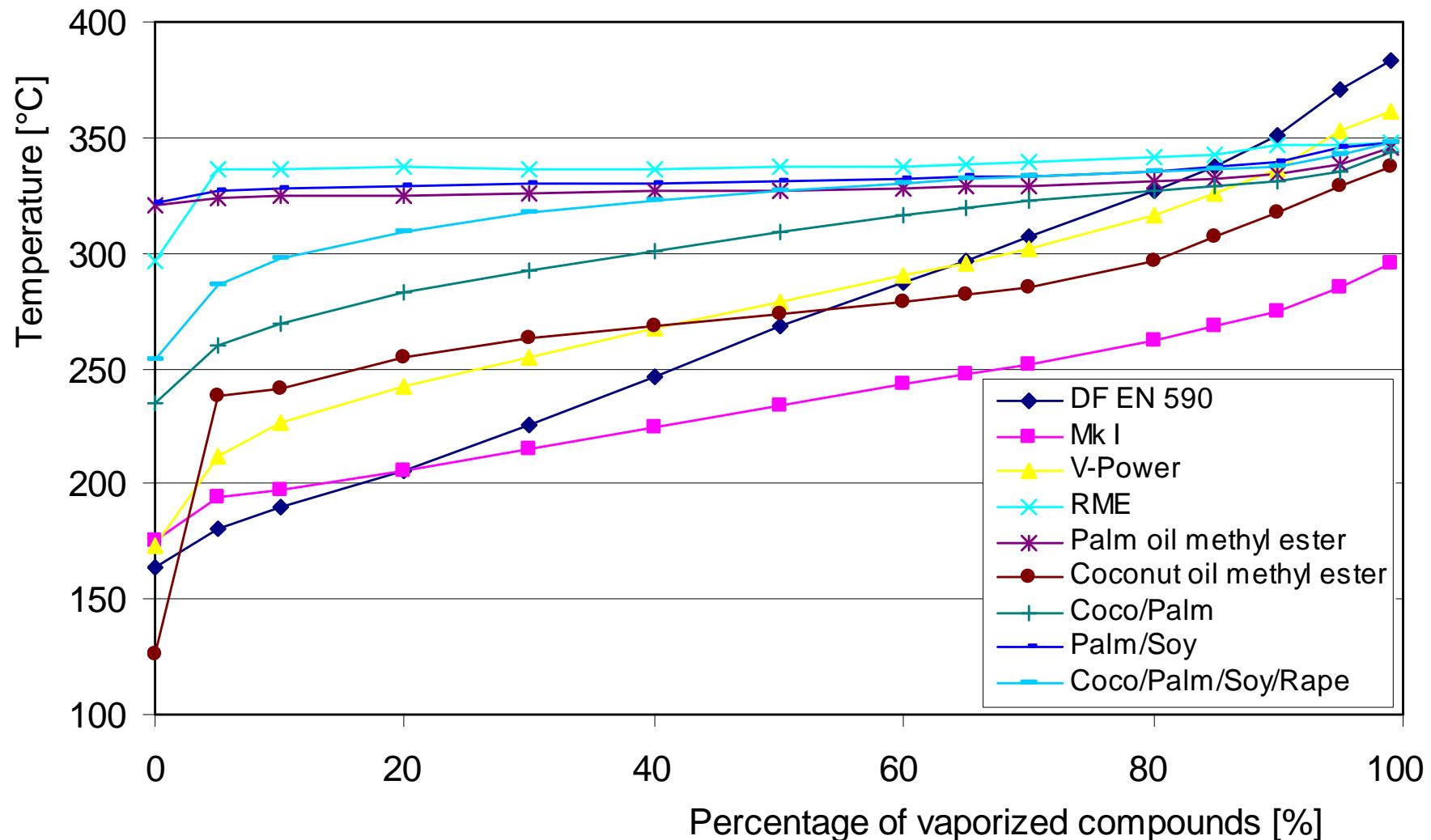
Modifications of FAME

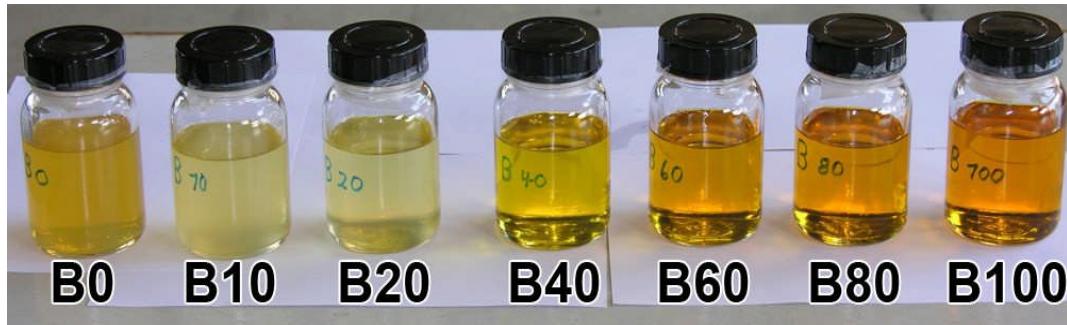
- Collection of some problems with neat biodiesel and blends
- Opportunities to solve the problems with blends
- Opportunities to solve the problems with neat biodiesel

Modifications of FAME

- Collection of some problems with neat biodiesel and blends
- Opportunities to solve the problems with blends
- Opportunities to solve the problems with neat biodiesel

Boiling Curves for Different Diesel Fuels



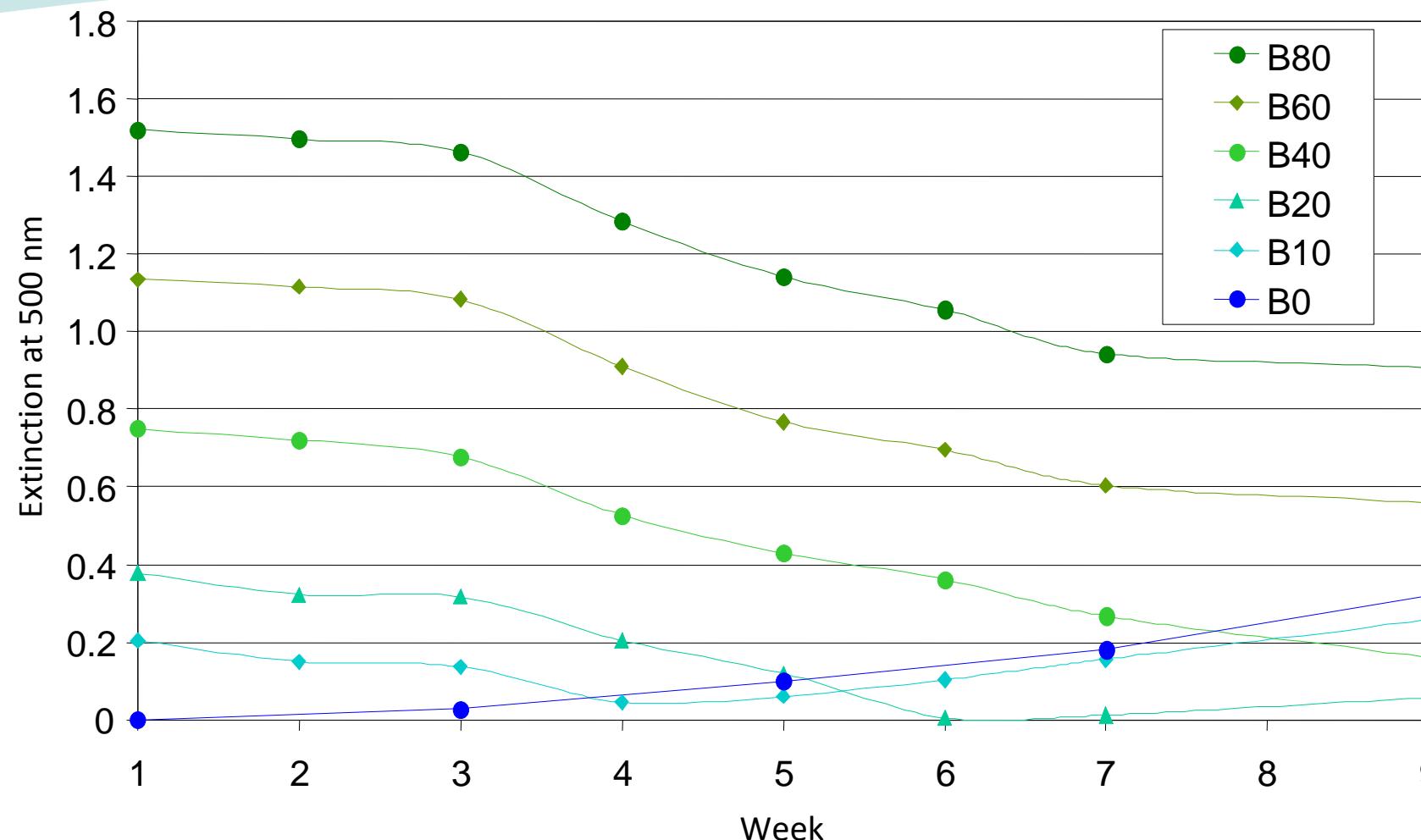


- Ageing experiments
- B0 ... B100

- B20, aged through 2 years

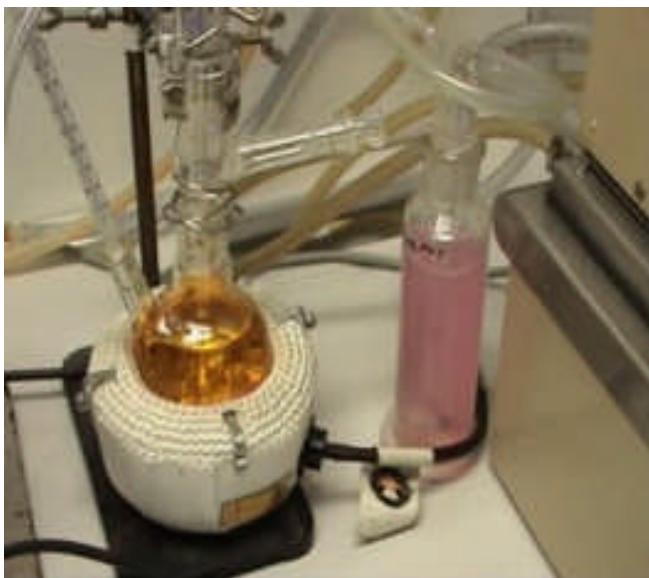


Spectroscopic Monitoring of Aging in the Sunlight



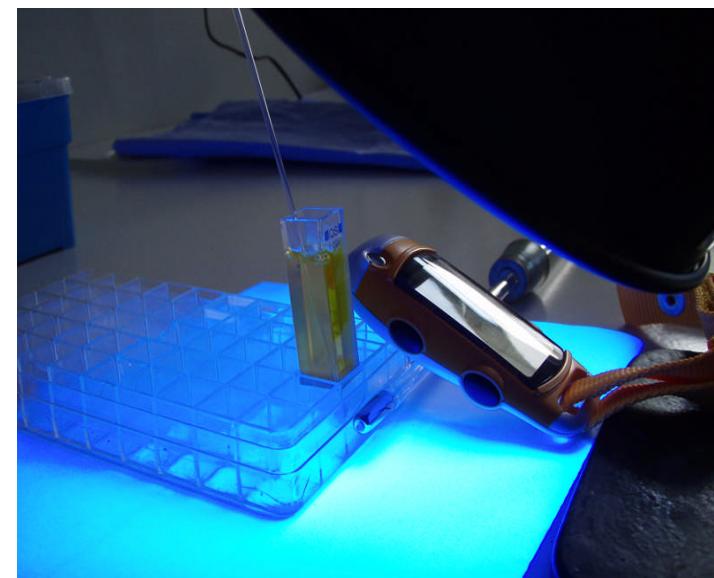
Thermo-oxidation

- $T = 110 \text{ } ^\circ\text{C}$
- Aeration

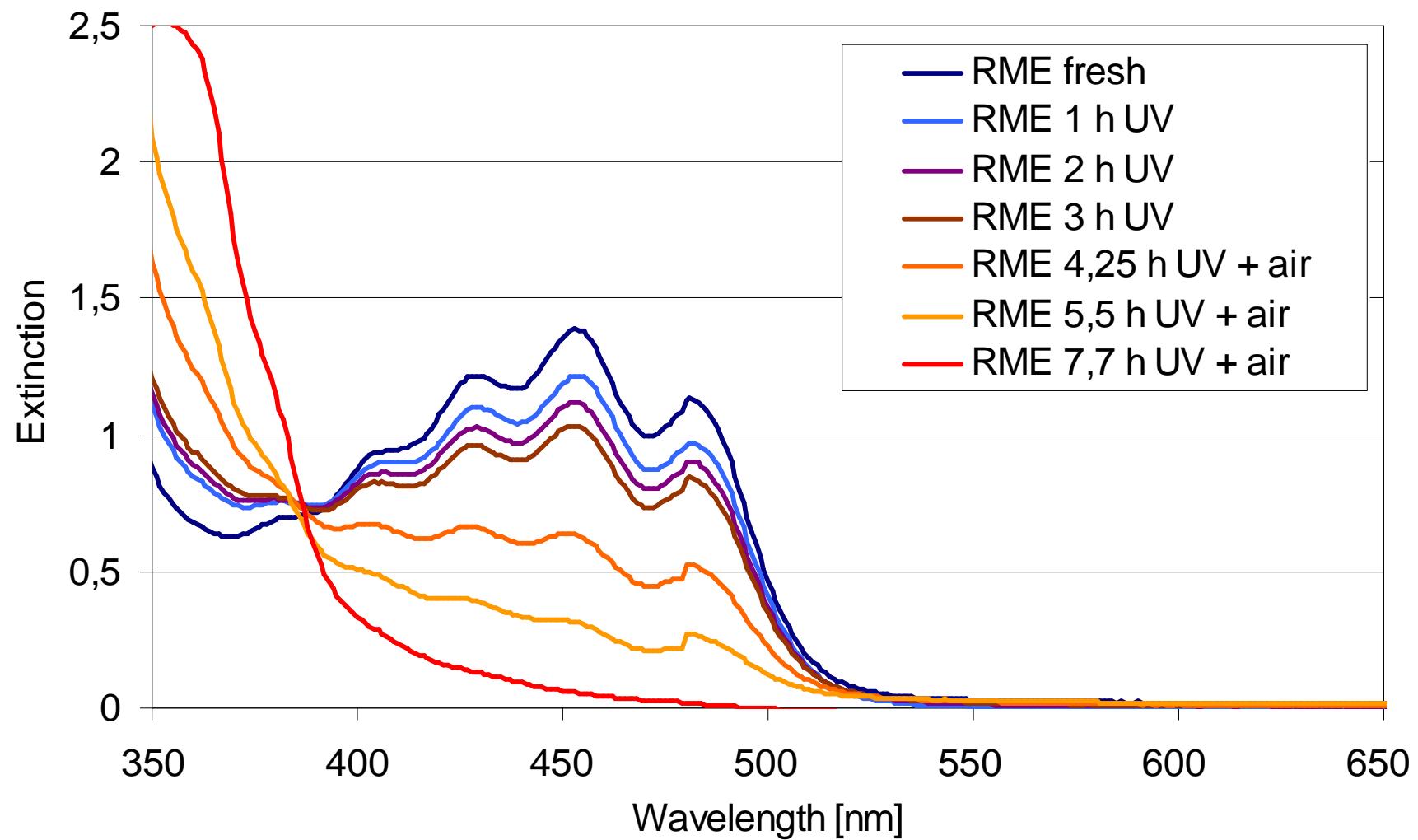


Oxidation by UV irradiation

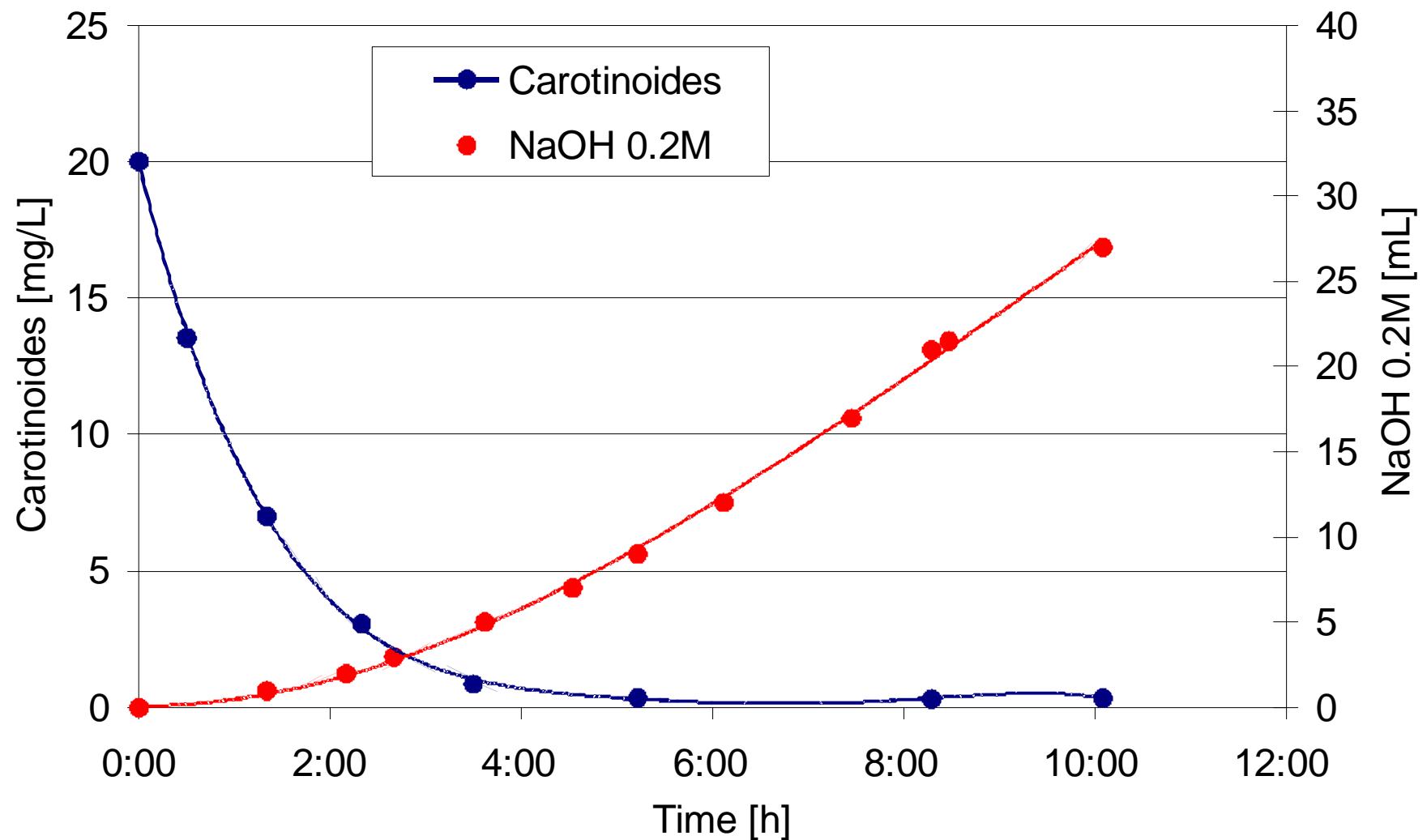
- Aeration
- $\lambda = 256 \text{ nm}$



Ageing through Oxidation by UV Irradiation



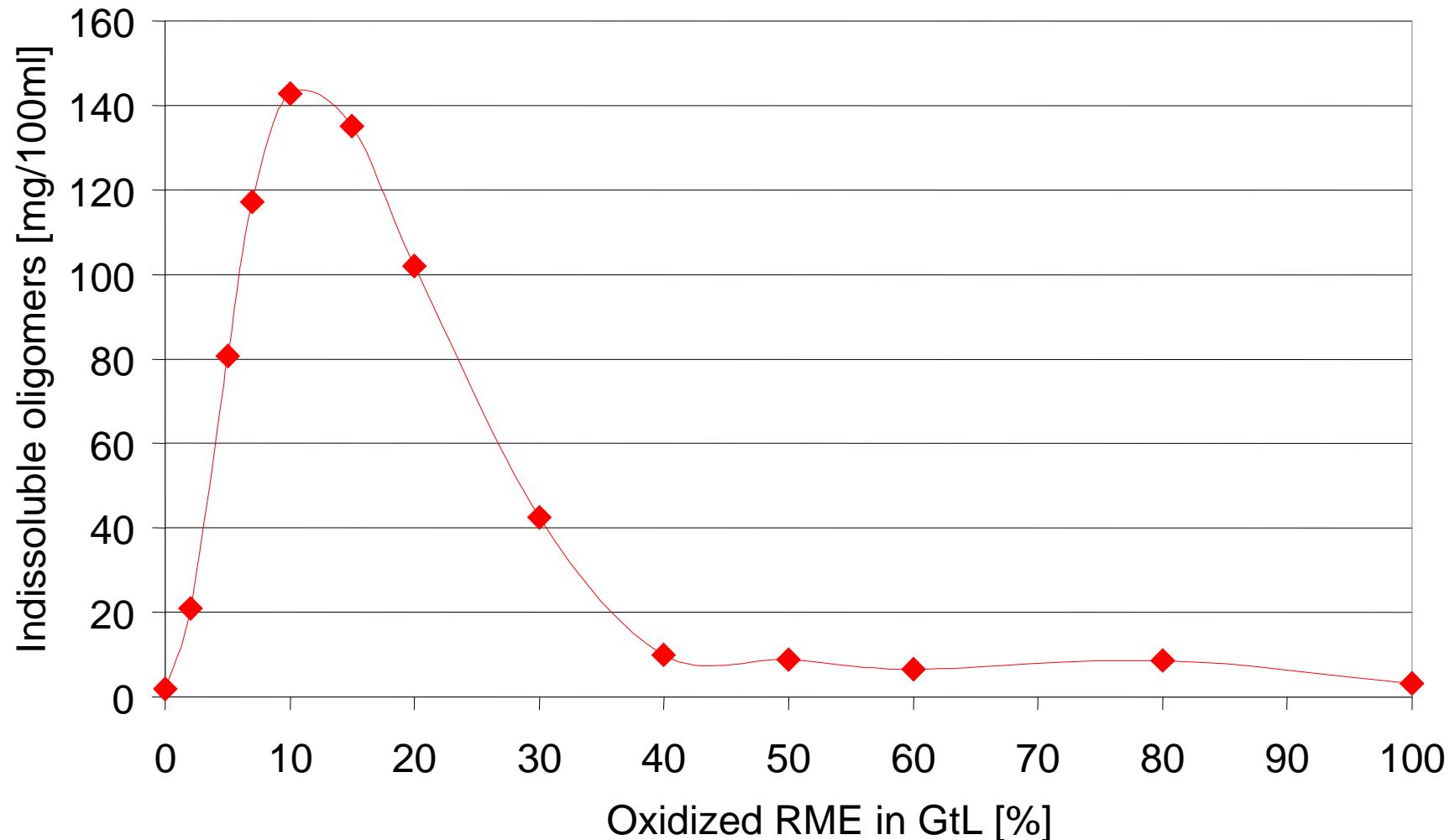
Thermo-Oxidation of RME Exposed to 140 °C in Air – Trends of Antioxidants and Acid Number



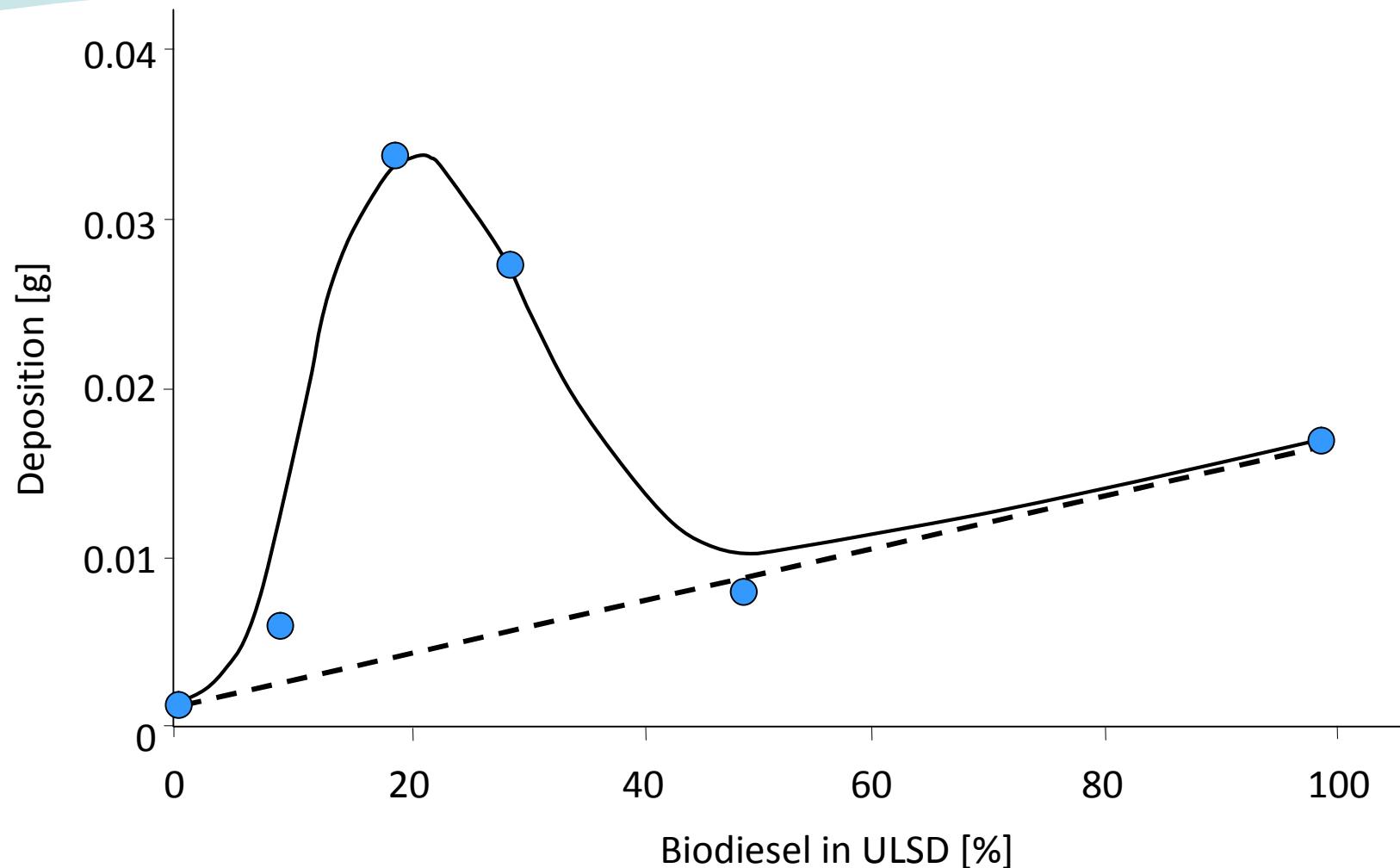
Sedimentation Occurs when Oxidized RME is Mixed with GTL



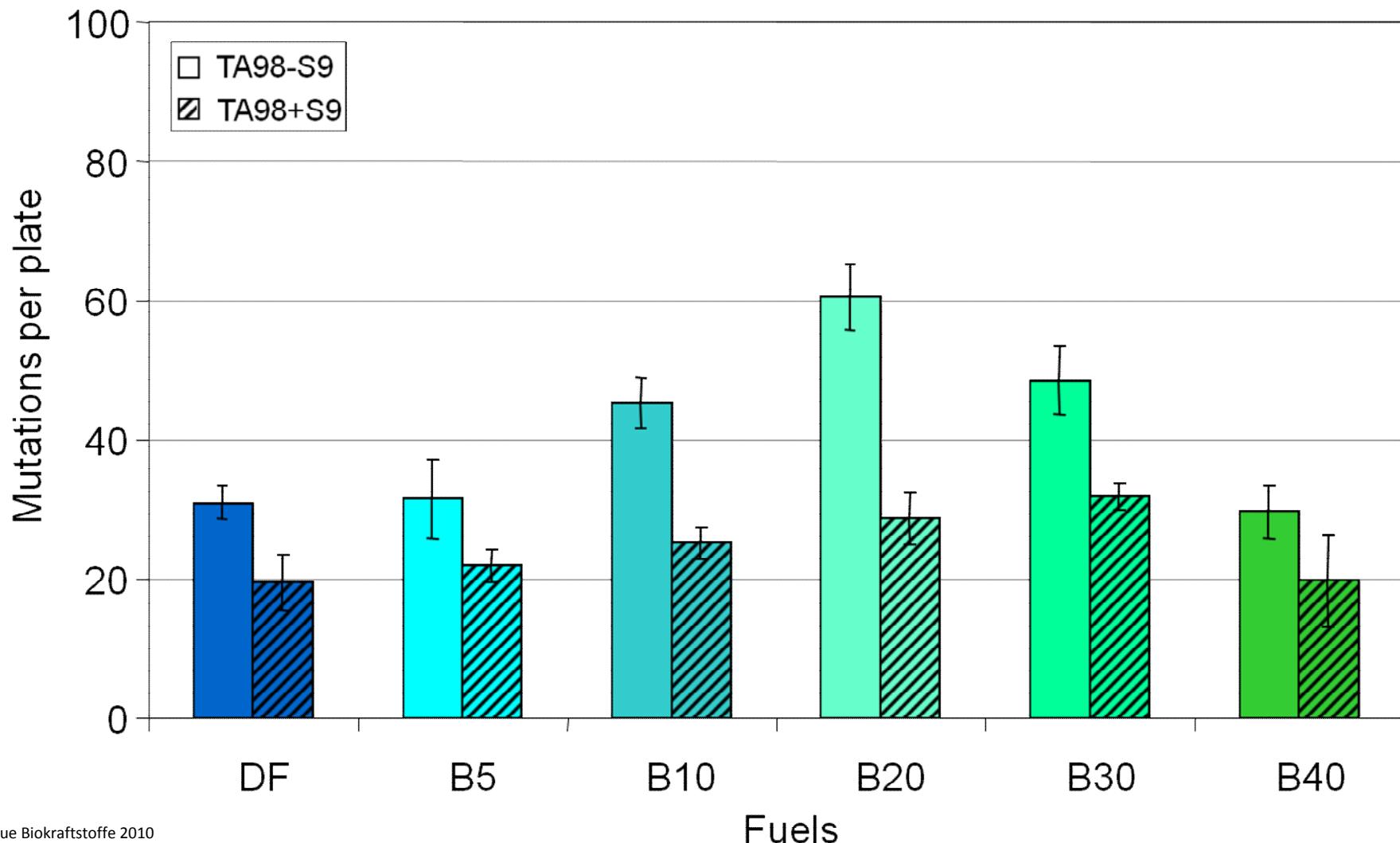
Indissoluble Oligomers in Blends of GtL and Oxidized RME



Sediment Formation in DF-Biodiesel Blends

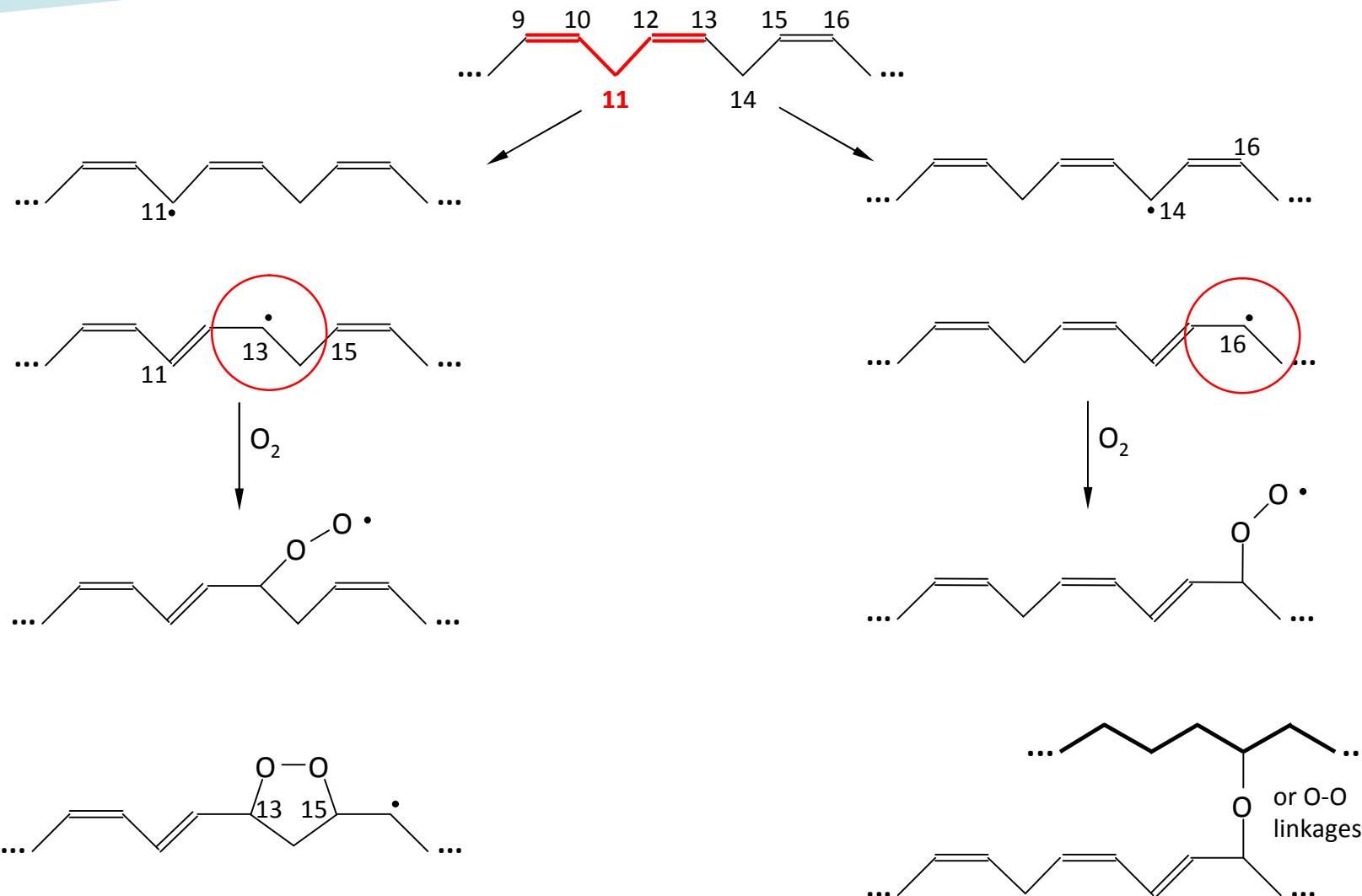


Mutagenicity of Particulate Extracts in the AMES Test; MAN D08, ETC Test

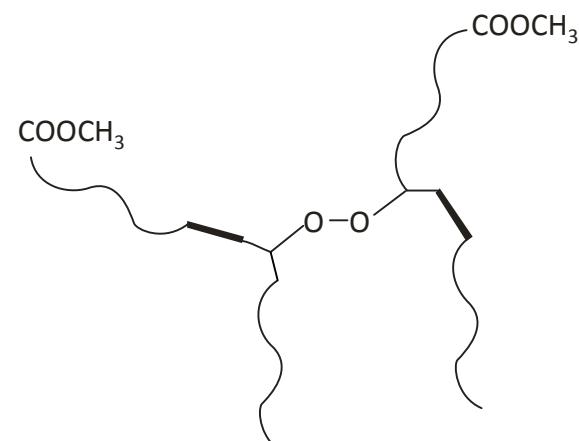


	Oligomers [% wt]	Biodiesel* [% wt]
Carbon	65.6	76.99
Hydrogen	9.6	12.27
Nitrogen	0.1	0
Oxygen	24.8	10.6
Sulphur	0	0

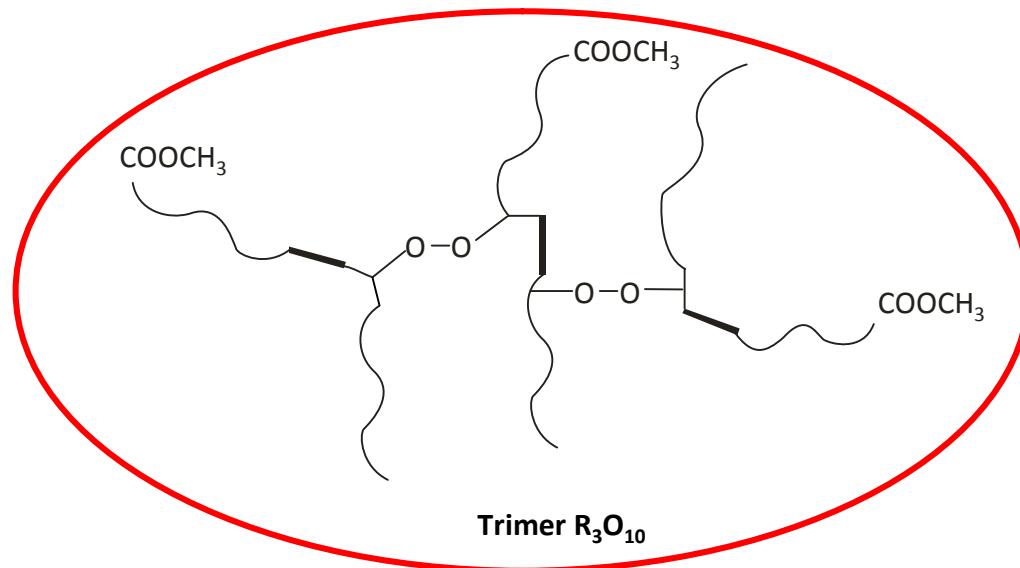
Possible Mechanism: Oligomerization



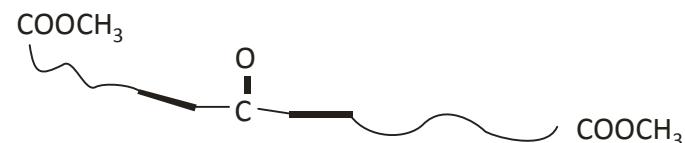
Possible Oligomers



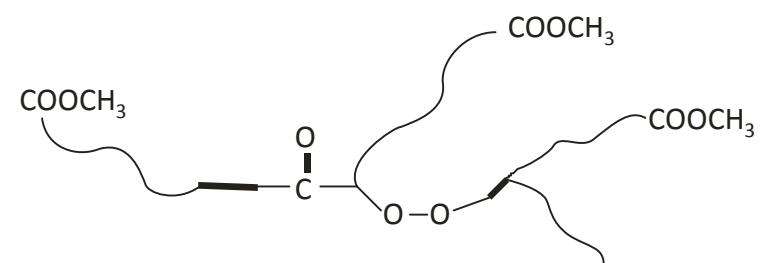
Dimer R_2O_6



Trimer R_3O_{10}



Aldol Dimer $\text{R}'_2\text{O}_5$ ($\text{R}' < \text{R}$)



Aldol Oligomer $\text{R}'_2\text{RO}_9$

Further Results from the Literature

Precipitate gels were detected by Peyton, McGinnis and Bureman (NALCO), too. These occurred in soy oil methyl ester and mixed FAME in the case of B5 and B20 – not in case of B100. No sedimentation was found for palm oil methyl ester in every blend (B5 ... B100).

For soy oil methyl ester and mixed FAME the precipitate forming compounds were also found in B100; however, in the neat fuels they stayed dissolved and did not precipitate.

The compounds were identified as azelaic acid monomethyl ester (formed by oxidative degradation of methyl oleate) and hexanoic acid (formed by the oxidative degradation of methyl linoleate).

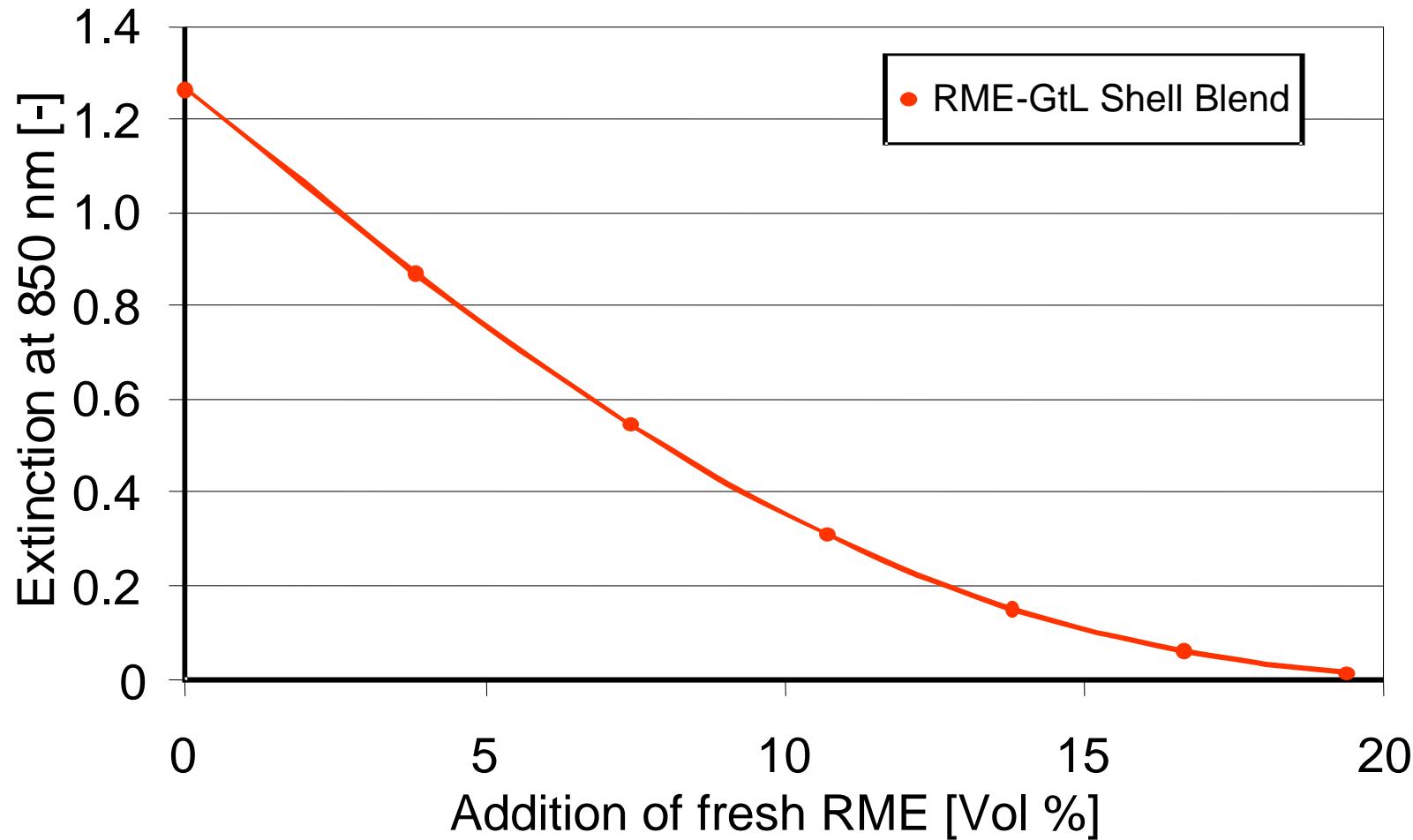
Modifications of FAME

- Collection of some problems with neat biodiesel and blends
- Opportunities to solve the problems with blends
- Opportunities to solve the problems with neat biodiesel

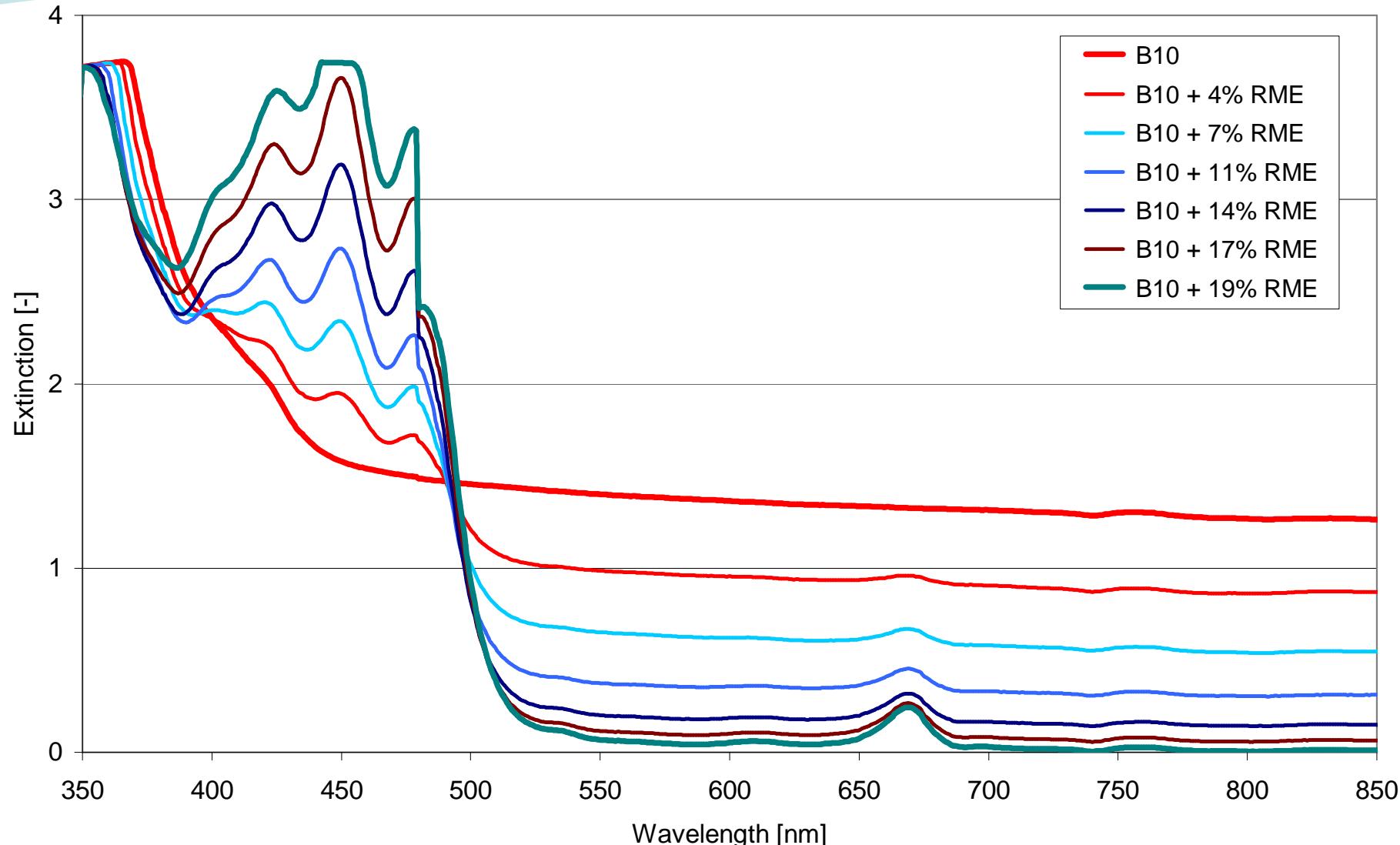
Dissolution of the Sediments



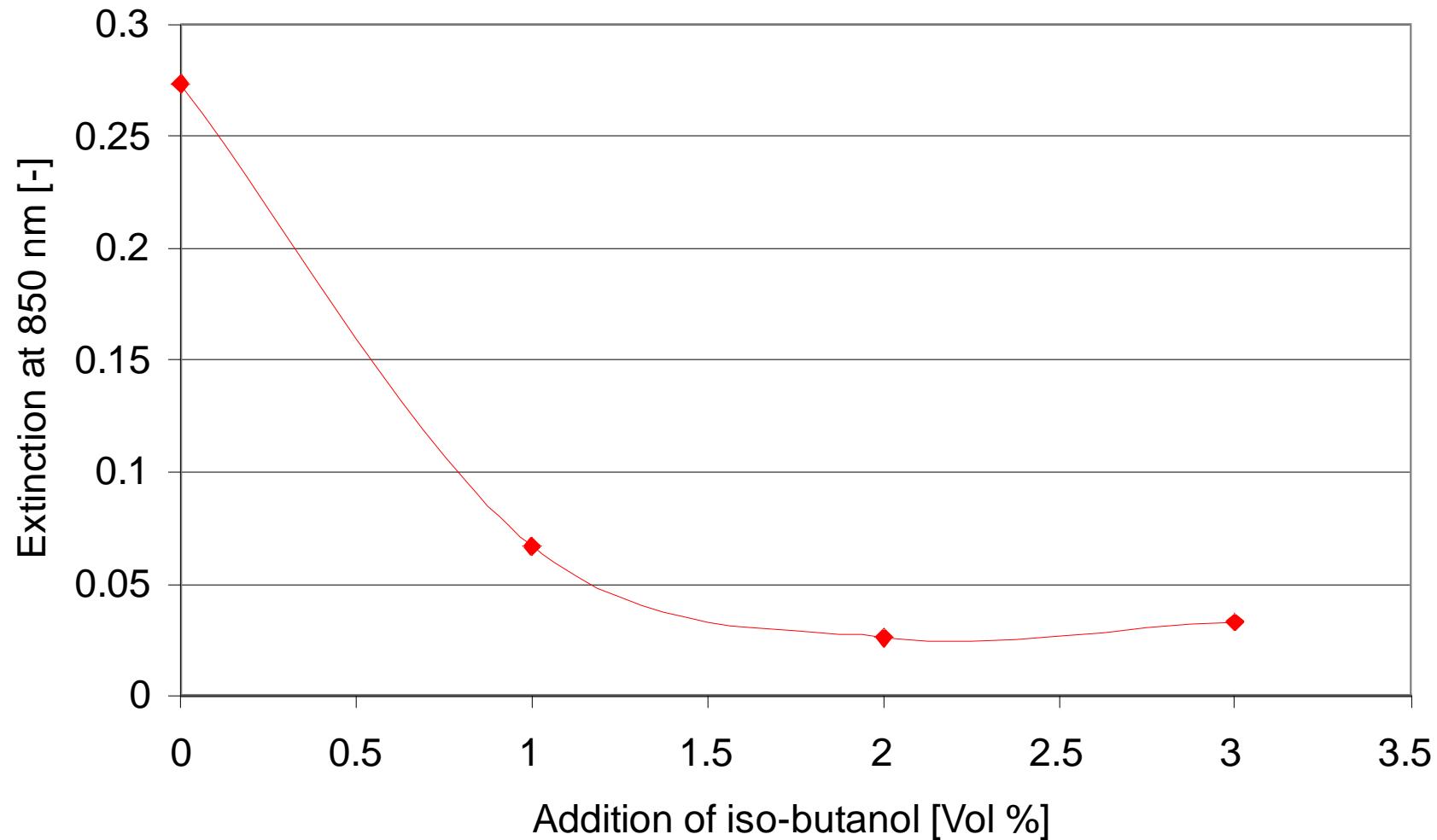
Dissolution of Sediments in B10GTL via Addition of Fresh RME (UV/VIS)



Dissolution of Sediments in B10GTL via Addition of Fresh RME (UV/VIS)



Dissolution of Sediments via Addition of Alcohols



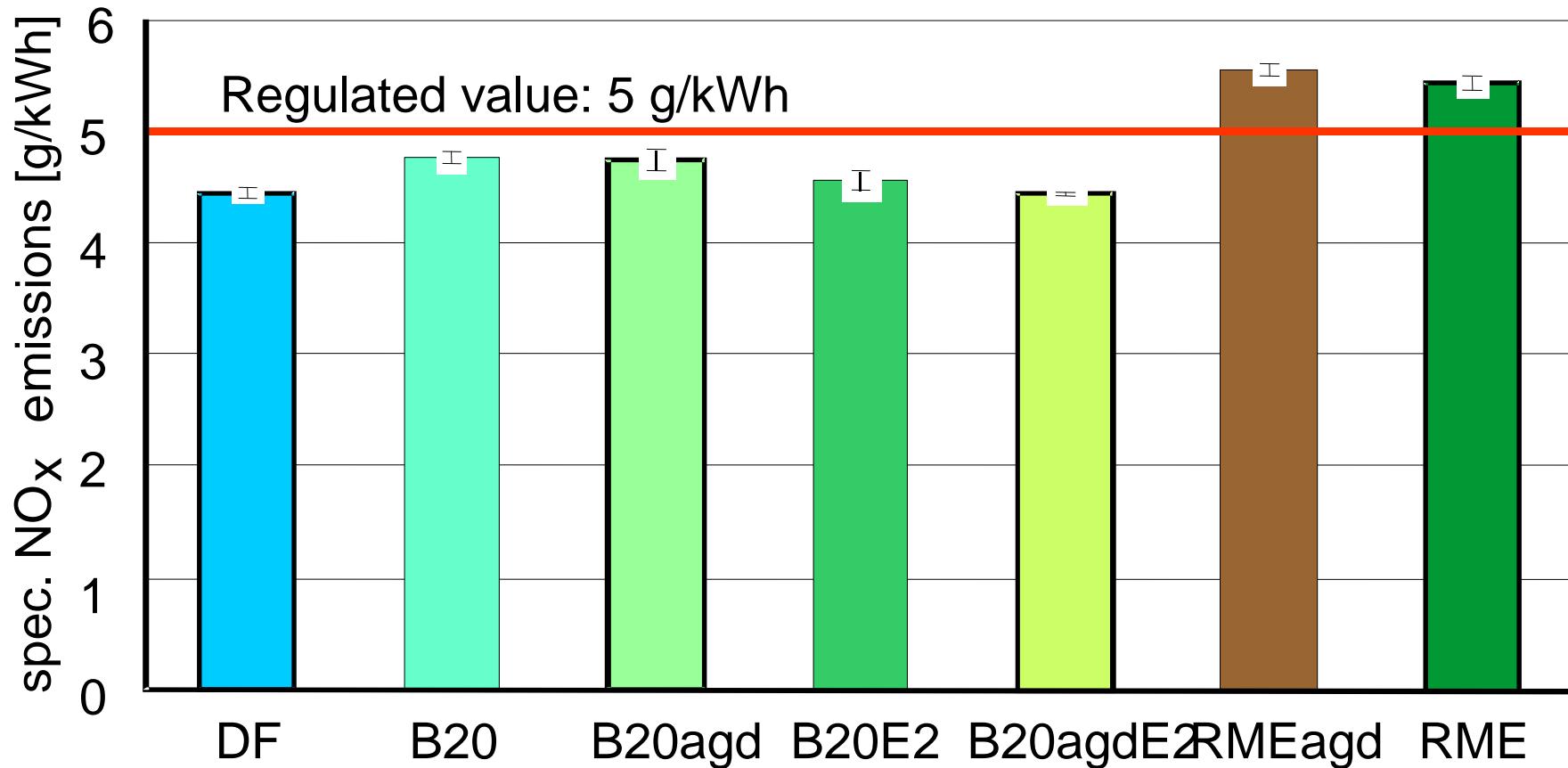
A dissolution of the sediments is achievable via addition of

- ethanol
- butanol
- octanol

This can also be observed when carrying out the filter blocking tendency test IP 387.

An addition of n-butanol to an existing B10 blend proved to be more efficient than an addition of n-butanol to the FAME before the preparation of the B10 blend.

Dissolution of Sediments via Addition of Ethanol – Effects on the Emission of NO_x; OM906, ESC Test



More Results from the Literature

Peyton, McGinnis und Bureman propose to add antioxidants in order to prevent sediment formation.

Modifications of FAME

- Collection of some problems with neat biodiesel and blends
- Opportunities to solve the problems with blends
- Opportunities to solve the problems with neat biodiesel

Results of the Biodiesel Workshop, Wuhan, 2007

Plant oil composition

- Boiling (start) temperature must be brought down \Rightarrow chain lengths 12...16
 - Melting point (resp.: CFPP) should be low enough \Rightarrow non-saturated fatty acids
 - No polymerization \Rightarrow 1 or 2, not 3 double bonds
- \Rightarrow C12:1 ... C16:2

Processing

- Low phosphorus content

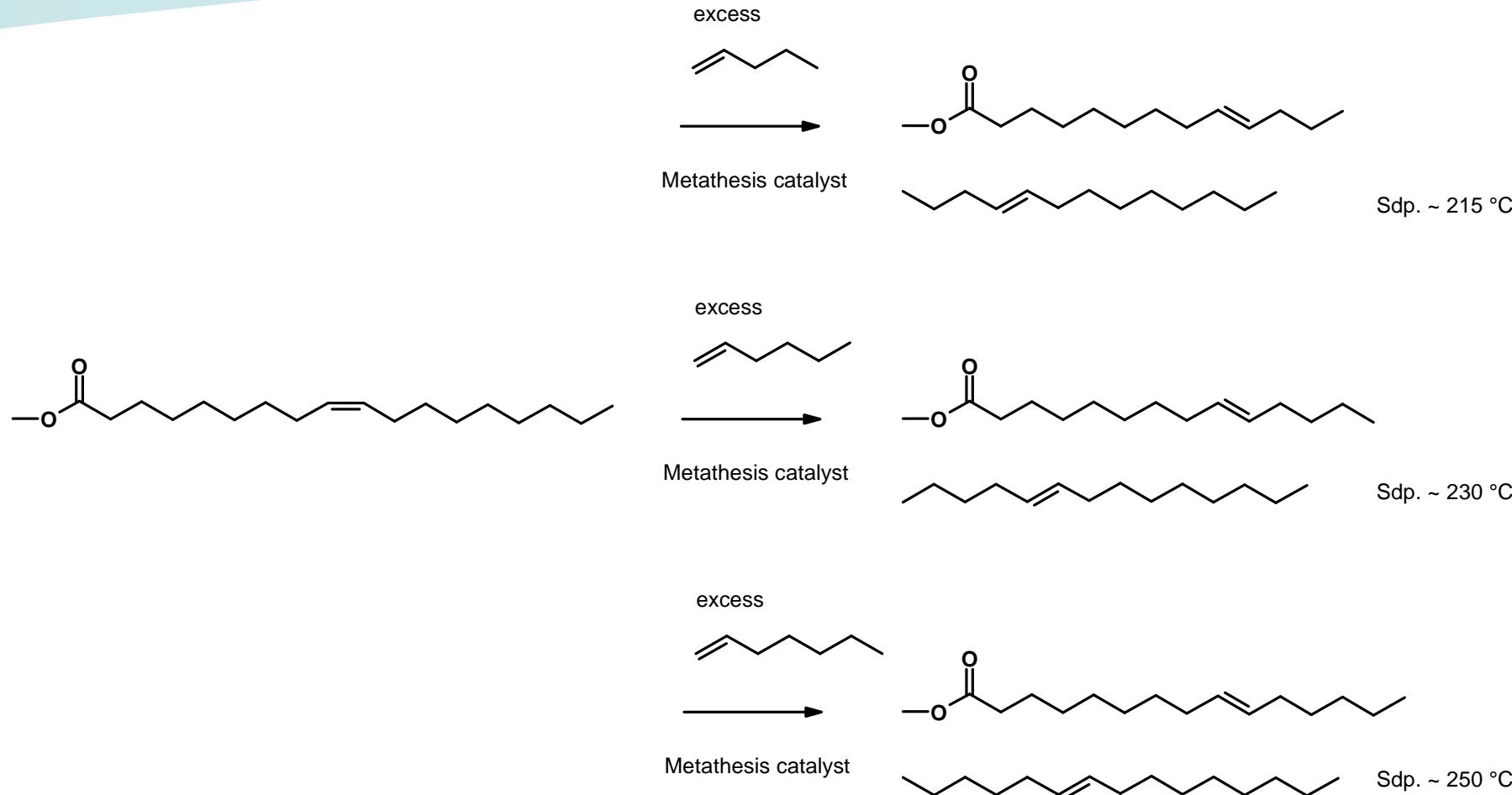
Environment and health

- Low summer smog formation potential
- Low mutagenic potency
- Low cytotoxic effects

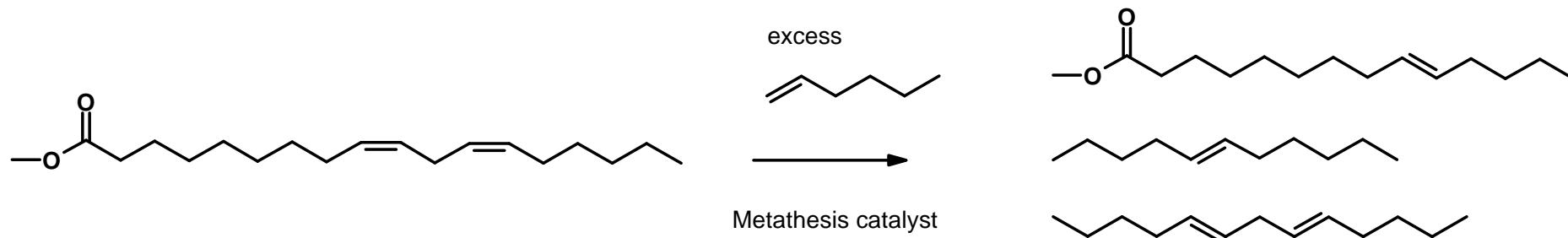
Further demands

- Biotechnical methods for transesterification?
- Alternative pathways for utilization of glycerol?

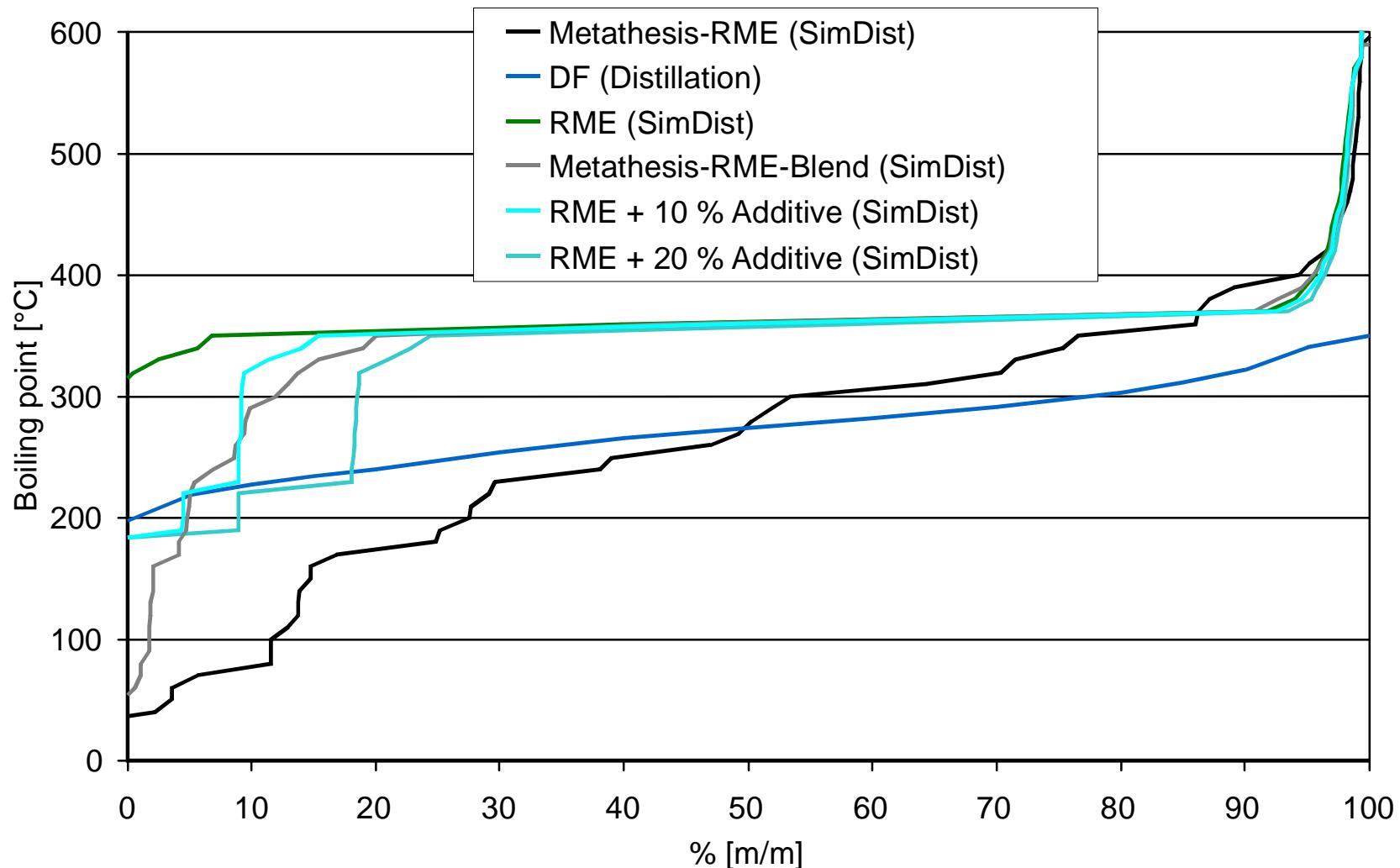
Schematic of the Cross-Metathesis of Oleic Acid Methyl Ester for the Formation of New Biodiesel Blends



Schematic of the Cross-Metathesis of Methyl Linoleate for the Formation of Biodiesel Compounds with Very Low Boiling Point



Influence of Metathesis Products and Additives on the Boiling Behaviour of Diesel Fuels



*Thank you very much for your kind
attention!*

Acknowledgement:

The authors thank the *Verband der Deutschen Biokraftstoffindustrie e.V.* (VDB) and the *Union zur Förderung von Oel- und Proteinpflanzen e.V.* (UFOP) for their support of the underlying research projects.